

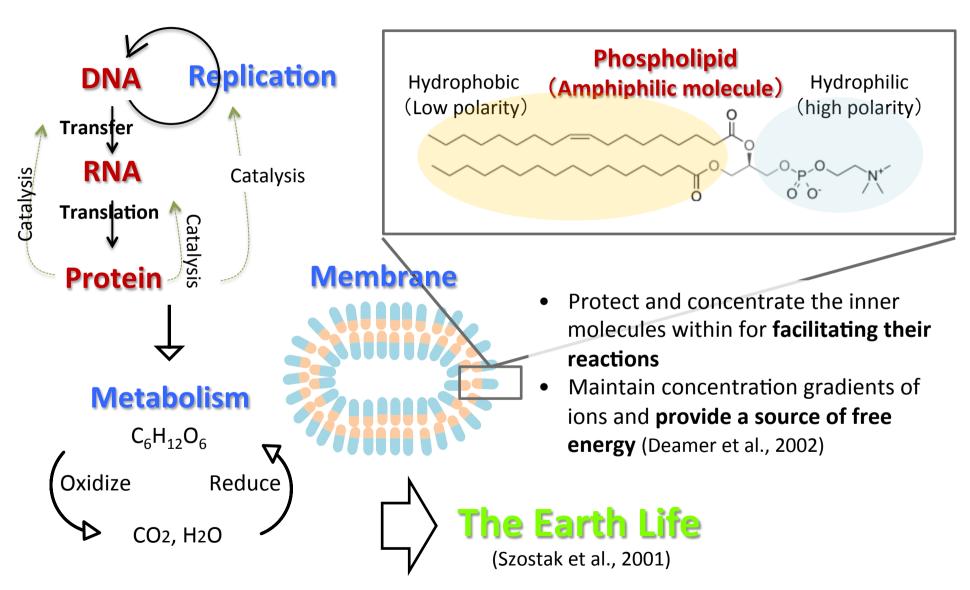
Dynamics observation of prebiotic microspheres through their chemical and morphological distributions:

Attempt to define "Planetary Life"

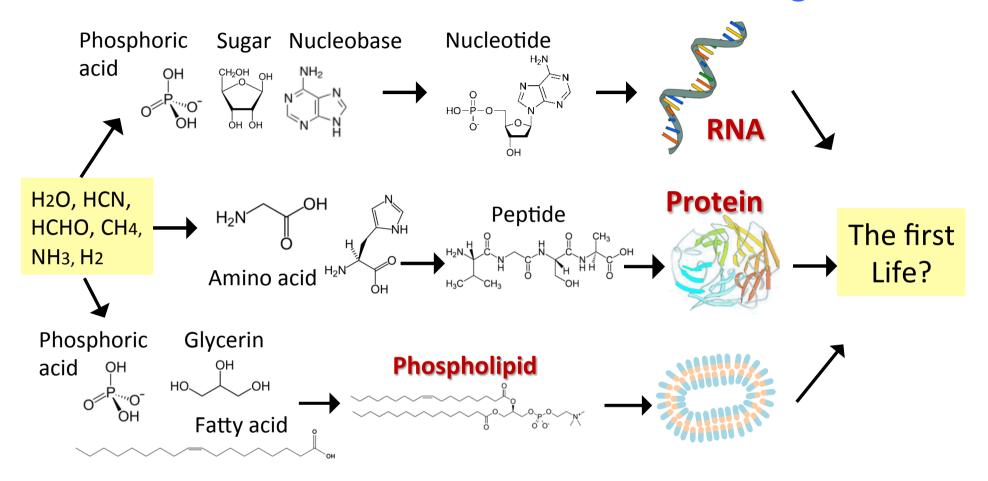
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Approach to Origins of Life based on the definition of "the present Earth-Life"



Traditional idea of chemical evolution toward Origin of Life

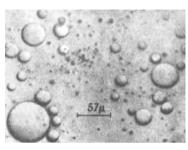


- Any prebiotic synthesis has not been able to demonstrate the formation pathways above (for more than 60 years since Millar's experiment!)
- Which came first, DNA or protein? ("Chicken or the egg" problem) is still unknown

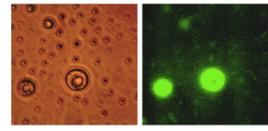
Difficult to model "the present" biomolecules

Prebiotic formation of organic microspheres

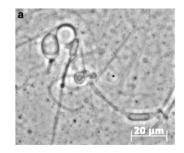
- ✓ Coacervate formed from gelatin and gum arabic (Oparin et al.,1976)
- ✓ **Protenoid microspheres** produced by thermal copolymerization of glutamic acid or aspartic acid (Fox and Harada ,1958)
- ✓ Microspheres formed from organic extracts of meteorites and interstellar ice analogue
 (Deamer, 1985; Deamer and Pashley, 1989; Dworkin et al., 2001)
- ✓ Acylglycerols synthesized by condensation of fatty acid and glycerol under hydrothermal condition (Simoneit et al.,2007)
- ✓ **Microspheres** formed from a mixture of polycyclic aromatic hydrocarbon (PAHs) and fatty acids (Groen et al.,2012)



(Fox and Harada ,1958)



— 30 μm (Dworkin et al., 2001)



(Groen et al.,2012)

However, the cell-like functions are not observed from the materials

Trying to identify <u>a boundary</u> between abiotic materials and bio-functions

Goal in this study

In order to <u>seek the protocell-like functions</u>, we investigated the <u>formation mechanisms of prebiotic organic microspheres</u> produced by Maillard-type reaction of formaldehyde (HCHO) and ammonia (NH₃) which could have been the common molecules in the early Earth.

◆ Maillard type reaction (Weber, 2001 modified)
(Browning reaction, Aminocarbonyl reaction)











Paraformaldehyde (HCHO) (2M) Glycolaldehyde (1M) NH4OH (0.5M) Ca(OH)2 (0.2M) H2O (2ml) (Cody et al., 2009)

Black participate

after 20 days→

Centrifuge

Ca²⁺ is removed by 2M HCl

Black insoluble residue

ultramicrotomed (100~300 nm)

Experimental

Optical microscope

Size, Textures

Scanning electron microscope (SEM)

Fourier transform infrared spectroscopy (FTIR)-attenuated total reflection (ATR)

Bulk compositions

Synchrotron-based microspectroscopy Local

compositions

 Scanning transmission X-ray microscope (STXM)-X-ray absorption near edge structure (XANES)

Observation of solid by optical microscope: Size distribution

5 days (liquid)

5 days (solid)

10 days (solid)

Sorry, the results are unpublished data.

23 days (solid)

51 days (solid)

81 days (solid)

Most of the microspheres were distributed inside of the organic solid. The size of the microspheres increased until 51 days, but <u>after that, smaller microspheres</u> <u>increased (81 days).</u>

Observation of solid by optical microscope: Morphology

5 days 10 days 51 days



Sorry, the results are unpublished data.



51 days



63 days 81 days

Observation of solid by scanning electron microscope (SEM)

10days (irregular shape)

20days (smooth surface)

51days (hollow)

Sorry, the results are unpublished data.

60days (constricted)

5 days

Size distributions of microspheres

10 days

Sorry, the results are unpublished data.

51 days

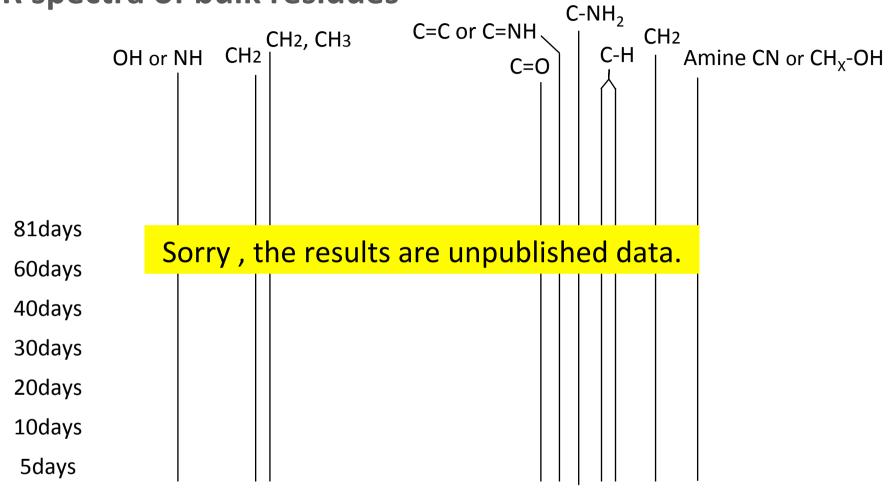
63 days

81 days

- Production of large microspheres is resulted from combination of small microspheres
- Re-production of smaller microspheres is due to **division** of larger microspheres

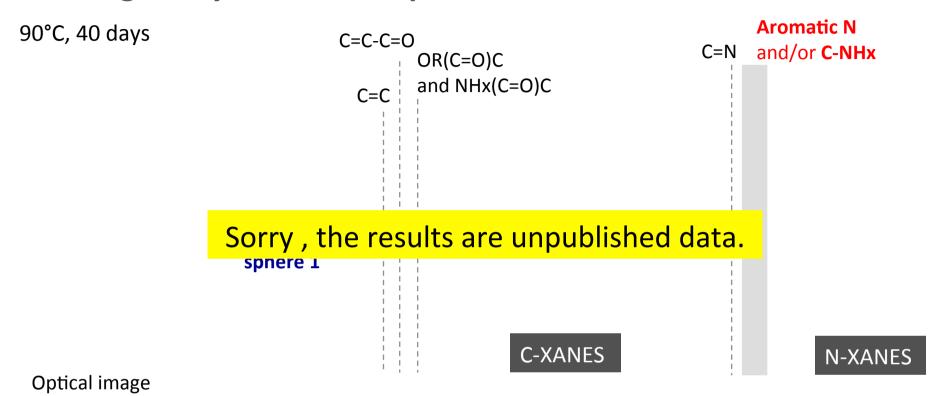
サイズ (μm)

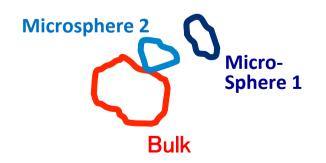
FTIR spectra of bulk residues



The functional group compositions were largely constant over heating time. The microspheres were formed by self-assembly of molecules due to different polarities.

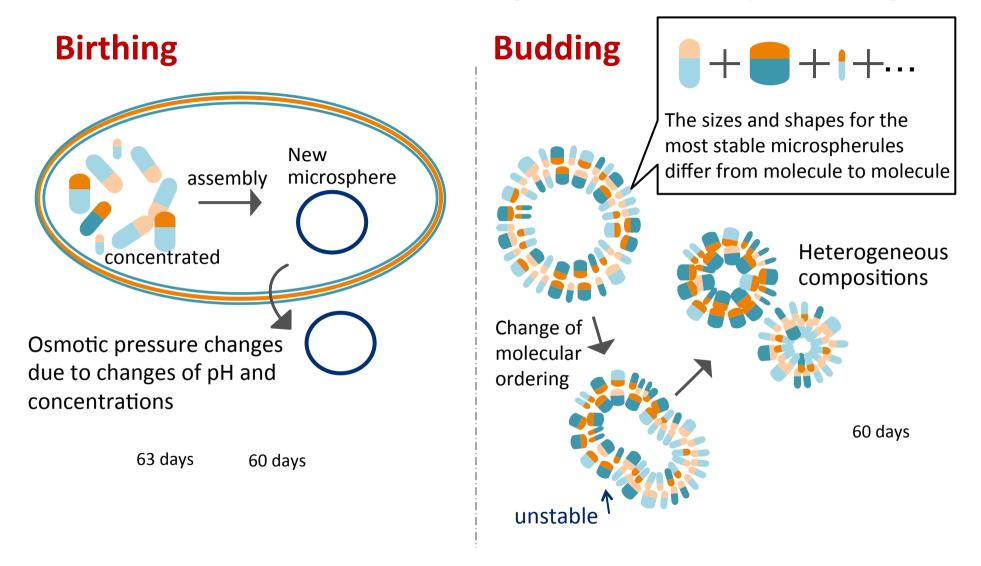
Heterogeneity of local compositions (STXM-XANES)





- ✓ Microspheres and host organics consist of the similar chemical compositions in most cases
- ✓ <u>Individual microspheres</u> have a slight extent of <u>chemical heterogeneities</u>

Division mechanisms of microspheres: Self-reproducing?



(Sakuma & Imai, 2011; Sakuma, et al., 2010 modified)

We suggest that this cell-like behavior of prebiotic microspheres linked to <u>self-replication</u>, as a definition of "Planetary Life" (which is beyond that of the Earth-life) (wishful thinking)

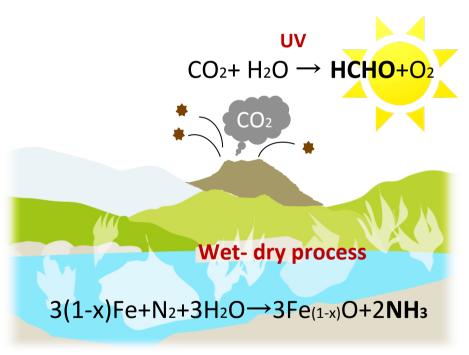
Plausible conditions for the growth and division of microspheres:

♦ Volcanic or Geothermal fields

- <u>Concentration of materials</u> by wetdry cycle
- <u>Continuous supply of materials</u> via photochemical reaction in atmosphere and mineral catalyzed synthesis (Washington, 1999)

Subduction zone

- <u>Concentration</u> and reactions of dissolved materials by dehydration (Holm and Neubeck, 2009)



These environments could exist not only the Earth but other planets such as Mars and Icy moons

Summary

- (1) Origins of Life is <u>NOT</u> necessarily <u>the so-called biomolecules</u> or <u>high-order functional molecules</u> of the <u>present Earth-life</u>.
- (2) Prebiotic organic microspheres synthesized in this study were formed by <u>self-assembly of the complex molecules</u>. They demonstrated <u>a self-reproducing-like behavior</u> by continuous growth and <u>division</u>.
- (3) The starting materials for formation of microspheres <u>do not</u> need to be <u>ammonia</u> (NH₃) and formaldehyde (HCHO), but their reaction products need to be <u>amphiphilic</u> (different polarity)
- (4) Plausible conditions for the growth and division of microspheres could be volcanic or geothermal field, or subduction zone, where materials are continuously supplied and concentrated.